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Chemistry is a branch of science which deals with the physical and chemical properties of substances, their composition and their usage in order to form new substances.

In order to study chemistry, every element that exists in nature and is known to us to date has been assigned a specific name and a one or two-letter symbol in order to get them handy for our own usage. Names of most of the elements are derived from English however their symbols are derived either from their English names or Latin names used for the elements.Example: Sodium is an element that is metal and its symbol 'Na' is derived from its Latin name 'Natrium'. Hydrogen is an element that is assigned an elemental symbol 'H' that is derived from its English name itself.Compounds are basically the combination of two or more elements that too in a specific ratio or in relative proportions. A chemical formula is an expression that represents the element in that compound along with its relative proportion in the compound. Example: Chemical formula for water is H2O which means that water is a compound that is formed by the combination of 2 proportions of hydrogen and 1 proportion of oxygen. Similarly, sulfuric acid has a chemical formula H2SO4 which means that in this compound, the relative proportion of H, S and O are: 2:1:4.Types of Chemistry Formulae or Chemical FormulaeThere are different types of chemical formulae. These are as follows.Molecular FormulaEmpirical formulaStructural formulaMolecular Formula- A molecular formula contains the chemical symbols of the constituent elements followed by numeric subscripts. The subscripts denote the number of atoms of the constituent elements. The molecular formula tells the exact number and type of atoms in a single molecule of a compound.Empirical Formula- The empirical formula represents the simplest whole-number ratio of all kinds of atoms present in a molecule of a compound. The molecular formula for a compound can be the same as or a multiple of the compound's empirical formula.Structural Formula- Molecular formulas are compact and easy to communicate but they do not tell about bonding and atomic arrangement and the bonding of atoms. For that, we need the structural formula. The structural formula indicates not only the number of atoms but also their arrangement in space.Example: Let us consider a glucose molecule and understand its molecular, empirical and structural formulae.The molecular formula for glucose is C6H12O6 which tells us the exact number of constituent atoms (carbon, hydrogen and oxygen written as C, H and O respectively) in one glucose molecule.The empirical formula of glucose is CH2O, which represents the whole number ratio of the constituent atoms, viz, C, H and O.The structural formula of glucose will indicate how each carbon, hydrogen and oxygen atom is bonded in a molecule of glucose. The structure of a glucose molecule is given below.

Image: The structural formula of glucose How to write a chemical formulaAs stated above, for an ionic compound, in general, cations are written first followed by the anion. For covalent compounds, the sequence-specific to the bonding is followed.For writing the formula for an ionic compound, the charges over the anionic and the cationic parts are crossed over in subscript. Such as on combining, H+ and PO4-3, the charges are crossed over to form H3PO4.One more example is given as the formation of Lithium oxide. Here Lithium-ion with +1 charge crossed over -2 charges on oxygen atoms to form Li2O. Image: Writing of chemical formula of Li2OList of Chemical Compound FormulaThe list of different types of chemical compounds is given below. Acid NameFormulaSulfurous acidH2SO3Sulfuric acidH2SO4Hyposulfurous AcidH2SO2Hydrochloric AcidHClNitrous acidHNO2Nitric acidHNO3Pernitric acidHNO4Carbonic acidH2CO3Oxalic acidH2C2O4Acetic acidCH3COOHPhosphoric acidH3PO4Phosphorous acidH2PO3Bromic acidHBrO3Hydroiodic acidHIIHydrofluoric acidHFCFormic acidHCOOHName of BasesFormulaLithium hydroxideLiOHSodium hydroxideNaOHPotassium hydroxideKOHCalcium hydroxideCa(OH)2Barium hydroxideBa(OH)2Aluminium hydroxideAl(OH)3Arsenic hydroxideAs(OH)3Ammonium hydroxideNH4OHPb(IV) hydroxidePb(OH)4Mercury(II) hydroxideHg(OH)2Tin(II) hydroxideSn(OH)2Common Salts: Different types of common salts are given below with examples.Acidic Salt: The salts which are formed by the partial neutralization reaction between a relatively strong acid and weak base and generally have ionisable H+ ions. Common ones are NaHSO4, KH2PO4.Basic Salt: Salt is formed by the reaction between a weak acid and a strong base. A common one is 2PbCO3. Pb(OH)2Neutral Salt: The salt that is formed by the reaction between a strong acid and a strong base. They are the combination of a cation and an anion. Example of neutral salt are NaCl, KCl etc. In general, the cation is written first followed by anion with their charges crossed to subscript. And using the table below, chemical formulae of several salts can be written.Common Cations and Anions are:CationsAmionsAmmonium NH4+Acetate CH3COO- Calcium Ca+2Carbonate CO3-2Iron Fe+3, Fe+2Chloride Cl- Potassium K+ Cyanide CN- Sodium Na+ Fluoride F- Copper Cu+ Nitrate NO3- Zinc Zn+ Nitrite NO2- Magnesium Mg+2Oxide O2- Lead Pb+2Phosphate PO4-3 Silver Ag+ Sulfate SO4-2Double Salt: Salts having more than one cation or anion.Common are: Mohr's salt - (NH4)2 Fe (SO4)2.6H2O. Potassium Alum - K2 SO4 Al2 (SO4)3.24H2O.Mixed salts: Salts having 2 salts in a fixed ratio.CaOCl2The chemical formula of some other commonly used compounds in chemistry:Glucose - C6H12O6Sugar/Sucrose - C12H22O11 Oxygen - O2Carbon dioxide - CO2Ammonia - NH3Ethanol (Alcohol) - C2H5OHPinegar - C2H4O2Baking soda - NaHCO3Hydrogen peroxide - H2O2Benzene - C6H6ConclusionSo, the chemical formula in chemistry is like a shorthand representation of the elements in a compound using the specific symbols assigned to each element that is mentioned in the periodic table. Chemists use symbols and formulae to represent elements, ions and compounds. Chemical equations model the changes that happen in chemical reactions. The given below table consists of the common names of important chemical compounds, their chemical formula and uses. The list has been prepared after a thorough analysis of previous year papers in which questions related to this topic were asked. This list is very helpful for the candidates preparing for various government exams like SSC CGL, State PCS, SSC CHSL, RRB NTPC, SSC CPO, RRB JE, SSC JE, etc. Read this study material and boost your preparation for the General Awareness section. Common Name Chemical Compounds/Elements Chemical Formula Uses Baking Soda Sodium Hydrogen Carbonate or Sodium bicarbonate NaHCO3 It is used as an ingredient in baking. Antiformin/ Bleach/ Chloride of Soda Sodium hypochlorite NaOCl/NaClO It is used on a large scale for surface purification, bleaching, odour removal and water disinfection. Bleaching Powder Calcium Oxychloride CaOCl2 It is used as a bleaching agent and disinfectant. Blue Vitriol Copper Sulphate CuSO4.xH2O It is used to control fungus and algae. Borax Sodium tetraborate decahydrate/ Sodium Borate Na2[B4O5(OH)4].8H2O It is used in laundry detergent. Calomel Mercurous chloride Hg2Cl2 It is used as a fungicide, and formerly in medicine as a purgative. Carboic Acid/ Hydroxybenzene/ Phenic Acid Phenol C6H5OH It is used as a disinfectant in household cleaners. Carborundum Silicon carbide SiC It is used in automotive brakes and clutches. Caustic potash Potassium hydroxide KOH It is used in liquid fertilizers, potassium soaps and detergents. Chile Saltpetre Sodium Nitrate NaNO3 It is used for making high-strength glasses, fertilizers and explosives. Chloroform Trichloromethane CHCl3 It is used as a solvent for lacquer, floor polishes and resins. Cream of tartar Potassium hydrogen tartrate/ potassium bitartrate KC4H5O6 It is used in bakery to stabilize egg whites. Dolomite Calcium magnesium carbonate CaMg(CO3)2 It is used as an ingredient in the production of glass, bricks, and ceramics. Dry Ice Solid carbon dioxide CO2 It is used to preserve food. Epsom salt Magnesium sulphate heptahydrate MgSO4.7H2O It is known to relieve pain and inflammation. Freon-12/CFC-12 Dichlorodifluoromethane CCl2F2 It is used as a refrigerant in refrigerators and air conditioners. Galena Lead (II) sulphide PbS It is also used in a green glaze applied to pottery. Glauber's Salt Sodium sulphate decahydrate Na2 SO4.10H2O It is used in medicine and as a dyeing agent. Glycerol Propane-1,2,3-triol C3H8O3 It is used in food and skin care industry. Green Vitriol Ferrous sulphate heptahydrate FeSO4. 7H2O It is used in sewage and water treatment, and as a pigment, fertilizer. Gypsum Calcium sulphate dihydrate CaSO4.2H2O It is used to manufacture dry wall. Common Name Chemical Compounds/Elements Chemical Formula Uses Heavy water Deuterium Oxide D2O It is used in nuclear reactors, as a neutron moderator. Hypo Solution Sodium Thiosulphate pentahydrate Na2S2O3.5H2O It is used for both film and photographic paper processing. Hematite iron (III) Oxide/ Ferric Oxide Fe2O3 It is used to produce pigments, preparations for heavy media separation, radiation shielding etc. Laughing gas Dinitrogen monoxide / Nitrous Oxide N2O It is used by dentist. Quick Lime/ Burnt Lime Calcium Oxide CaO It is used in production of iron and steel, paper and pulp and in treatment of water. Slaked Lime Calcium Hydroxide Ca(OH)2 It is used in mortars, plasters, and cements. Limestone Calcium carbonate CaCO3 It is used as a building material. Litharge(Red) and Massicot (Yellow) Lead (II) Oxide/ Lead Monoxide PbO It is used in the manufacture of Lead Stabilizers pottery, Lead glass, paints, enamels, and inks. Lye/ Caustic soda Sodium hydroxide NaOH It is used in manufacture of pulp and paper, alumina, soap and detergents. Marsh Gas Methane CH4 It is poisonous gas mainly found in mines. Methyl ethyl ketone/ Butanone Butan-2-one C4H8O It is mainly used in paint formulation. Milk of magnesia Magnesium hydroxide Mg(OH)2 It is used as an antacid to relieve indigestion, sour stomach, and heartburn. Mohr's Salt Ammonium iron (II) sulphate (NH4)2Fe(SO4)2 · 6H2O It is used to increase the shelf life of iron and reduce oxidation. Aqua Fortis/ Spirit of Niter Nitric acid HNO3 It is used for production of explosives, and fertilizers. Muriatic acid Hydrochloric acid HCl It is used as a food additive, and in gelatin production. Leather processing, etc. Oil of Vitriol Sulphuric acid H2SO4 It is use for making fertilizer. Wintergreen oil Methyl salicylate C8H8O3 It is used in traditional medicine for the bacterial infections. Phosgene Carbonyl dichloride COCl2 It is used to make plastics and pesticides. Plaster of Paris Calcium sulphate hemihydrate CaSO4.1/2H2O It is used as a protective coating on walls and ceilings. Potash Alum/Potassium Alum Potassium aluminium sulphate dodecahydrate KAl(SO4)2·12H2O It is commonly used in water purification, leather tanning, dyeing. Rochelle Salt Potassium sodium tartrate KNaC4H4O6.4H2O It is used in manufacturing of pectins and gelatines. Quartz Silicon dioxide (Silica) SiO2 It is used as oscillators in radios, watches. QuickSilver Mercury Hg It is used in thermometers, thermostats, barometers Table Salt/Common Salt Sodium chloride NaCl It is commonly used as a condiment and food preservative. 2-propanol/Propan-2-ol Isopropyl alcohol (CH3)2CHOH It is used to treat tremors, chest pain, hypertension, heart rhythm disorders, and other heart or circulatory conditions. Sal ammoniac Ammonium chloride NH4Cl It is used as cleaning agent in several industrial processes. Saltpetre Potassium nitrate KN03 It is used as a common food preservative. Slaked Lime Calcium hydroxide Ca(OH)2 It is used in form mortars, plasters, and cements. TNT Trinitrotoluene C6H2(NO2)3CH3 It is used in used in military shells, bombs, and grenades. Vermilion/ Cinnabar Mercuric sulphide HgS It is used as a pigment, and carve it into jewelry and ornaments. Washing Soda Sodium carbonate decahydrate Na2CO3. 10H2O It is used in commercial detergent mixtures to treat hard water. White Vitriol Zinc Sulphate ZnSO4.7H2O It is used for preserving skins and wood. Rs.449/- Read More Rs.398/- Read More Rs.199/- Read More Rs.199/- Read More

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